## 2,12'-BIS-HAMAZULENYL FROM Ajania fruticulosa ESSENTIAL OIL

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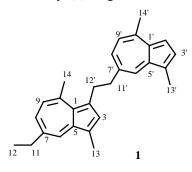
A new bisazulene, the structure of which was established using x-ray structure analysis and PMR and <sup>13</sup>C NMR spectra, was isolated from essential oil of Ajania fruticulosa obtained by steam distillation.

Key words: Ajania fruticulosa, 2,12'-bis-hamazulenyl, XSA, <sup>13</sup>C NMR, PMR.

The genus Ajania Poljak (Asteraceae) in Russia and adjacent states numbers 10 species [1], the most common of which is Ajania fruticulosa.

Known flavonoids and phenolcarboxylic acids [2, 3], a new germacranolide ajanolide [3], and three new guaianolides [4-6] have been observed in the aerial part of this plant. Steam distillation isolated the dark blue essential oil [7], the color of which is due to the presence of hamazulene [8], the content of which in the oil increases greatly with increasing time of distillation [7].

During chromatography of the essential oil over aluminum oxide with elution by petroleum ether, two dark blue bands form on the column. The upper of these contains hamazulene; the lower, as it turns out, corresponds to a new crystalline dark blue compound with mp 136-138°C. According to high-resolution mass spectrometry, it has empirical formula  $C_{28}H_{30}$ . It can be assumed from the color in combination with the empirical formula that the compound is some sort of azulene dimer. We resolved this issue by performing an x-ray structure analysis (XSA) and found that the compound is in fact a dimer of hamazulene and has the structure 2,12'-bis-hamazulenyl (1). Figure 1 shows the structure.



The azulene fragments of **1** are planar with a mean-square deviation of the atoms of 0.010 and 0.004 Å for C1-C10 and C1'-C10', respectively (Fig. 1). The interplanar angle between these fragments is  $2.3(5)^{\circ}$ . The ethyl bridge is out of the planes of the azulene fragments [torsion angles C3-C2-C12'-C11' -88(1)° and C12'-C11'-C7'-C8' 87(1)°]. The bonds lengths are the same within experimental uncertainty as those in the structurally similar synthetic dimers (*E*)-1,2-bis(3-guaiazulenyl)ethylene [8] and (1*R*,2*S*)-1,2-bis(4-(methoxycarbonyl)phenyl)-1,2-bis(3-guaiazulenyl)ethane [9]. The XSA of a specially prepared hamazulene derivative has been reported [10]. The molecules in the crystal are stacked along the *b* axis. However, the large (4.948 Å) distances between the centers of the molecules precludes a  $\pi$ -stacking interaction.

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Atom	C <sub>i</sub>	H <sub>i</sub>	Atom	C <sub>i</sub>	H <sub>i</sub>
1	133.14 s	-	1'	138.04 s	-
2	138.63 s	7.30 br.d $(J = 2.0)$	2'	113.67 d	7.34 d (J = 4.0)
3	140.57 d	7.43 s	3'	136.72 d	7.67 br.d $(J = 4.0)$
4	136.91 s*	-	4'	136.91 s*	-
5	124.29 s	-	5'	125.48 s	-
6	134.66 d	8.05  br.d (J = 2.0)	6'	135.33 d	8.13 br.d (J = 1.0)
7	134.54 s	-	7'	133.70 s	-
8	135.33 s	7.08 dd (J = 10.5; 2.0)	8'	136.95 d	7.25 dd (J = 10.5; 2)
9	126.26 d	6.64 d (J = 10.5)	9'	124.90 d	6.78 dd (J = 10.5)
10	145.07 s	-	10'	144.24 s	-
11	33.42 t	2.54 k (J = 8.0)	11'	44.80 t	3.17 distort. t $(J = 8)$
12	17.32 k	1.21 t (J = 8.0)	12'	35.56 t	3.61 distort. t (J = 8)
13	13.02 k	2.55 (3H) br.s**	13'	12.92 k	2.55 (3H) br.s**
14	26.97 k	2.75 (3H) br.s	14'	24.06 k	2.60 (3H) br.s

TABLE 1. <sup>13</sup>C NMR and PMR Spectra of 1 ( $C_6D_6$ ,  $\delta$ -scale, J/Hz, 0 = TMS)

\*Signals for C-4 and C-4' overlap, \*\*signals for 3H-13 and 3H-13' differ by about 0.004 ppm.

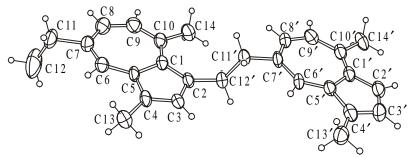


Fig. 1. Molecular structure of 2,12'-bis-hamazulenyl.

Signals in the PMR and <sup>13</sup>C NMR spectra of **1** (Table 1) were assigned using COSY <sup>1</sup>H—<sup>1</sup>H, <sup>13</sup>C—<sup>1</sup>H and COLOC <sup>13</sup>C—<sup>1</sup>H two-dimensional (2D) spectra. The most important observed cross peaks in the 2D COLOC spectrum were C-2/H-3, C-6/H-11, C-6/H-11', C-7/H-12, C-7'/H-11', C-9/H-14, C-9'/H-14', C-10/H-14, and C-10'/H-14'.

Natural guaiazulene and its biologically active dimers have also been found in marine coral of the genus *Gorgonaceae*, where they impart color to these marine organisms [11]. Dimeric guaianolids are encountered in plants and are possible precursors of dimeric hamazulene **1** and its analogs that have not yet been prepared [12].

## EXPERIMENTAL

IR spectra were recorded on a Vector 22 instrument. UV spectra were recorded on an HP 8453 spectrophotometer; NMR spectra, on a Bruker DRX-500 spectrometer (working frequency 500.13 MHz for <sup>1</sup>H, 125.76 MHz for <sup>13</sup>C). The high-resolution mass spectrum (EI, 70 eV) was obtained in a Finnigan MAT 8200 instrument.

Raw material of the leafy part of *Ajania fruticulose* (Ledeb.) Poljak growing near the village Dogolan of Eastern Kazakhstan District was collected during budding at the beginning of August 2004. Essential oil was obtained by steam distillation in a Clevenger apparatus over 3 h. The yield of oil was 0.70% calculated for air-dried raw material.

**Compound 1.** A weighed portion of essential oil (34.0 g) was chromatographed over neutral  $Al_2O_3$  with elution by petroleum ether. A colorless mixture of components eluted first and then hamazulene (dark blue oil, 12.1 g) and finally dark blue crystals of **1** (0.10 g), mp 136-138°C,  $C_{28}H_{30}$ .

UV spectrum (CHCl<sub>3</sub>, λ<sub>max</sub>, nm): 247, 293, 308, 355, 372, 657 (log ε 4.62, 4.79, 4.56, 4.04, 3.00, 2.12).

IR spectrum (KBr, v, cm<sup>-1</sup>): 2960, 2923, 2858 (C–H), 1546, 1520, 1422, 1368, 962, 864, 806, 773, 711.

Mass spectrum (EI, 70 eV, *m/z*, *I*<sub>rel</sub>, %): 366 (42) [M]<sup>+</sup>, 189 (67), 177 (100), 181 (32), 180 (15), 169 (23), 167 (42), 165 (27), 152 (24), 141 (15).

**X-ray structure analysis of 1** was carried out on a Bruker P4 diffractometer (Mo K $\alpha$ -radiation, graphite monochromator,  $\omega$ -scanning) at room temperature. Compound **1** crystallizes as very thin plates of poor quality. Therefore, data were collected in the range  $2\theta < 40^{\circ}$ . The crystallographic parameters are monoclinic, a = 12.988(5), b = 4.948(4), c = 33.466(12) Å,  $\beta = 94.47(2)^{\circ}$ , V = 2144.1(19) Å<sup>3</sup>, space group  $P2_1/n$ ,  $C_{28}H_{30}$ , Z = 4,  $D_c = 1.135$  g/cm<sup>3</sup>,  $\mu = 0.064$  mm<sup>-1</sup>, crystal size  $1.1 \times 0.3 \times 0.02$  mm. The structure was solved by direct methods using the program SHELXS-97 and refined by anisotropic and isotropic least-squares methods using the program SHELXS-97. Positions of H atoms were calculated geometrically. The final R factors were wR<sub>2</sub> = 0.4288 and S = 1.141 for 1985 independent reflections (R = 0.1268 for 1063 reflections with  $I > 2\sigma$ ). Crystallographic data for **1** and parameters of the x-ray experiment were deposited in the Cambridge Crystallographic Data Center (No. CCDC 299273).

## REFERENCES

- a) *Flora of the USSR* [in Russian], Vol. XXVI, Izd. Akad. Nauk SSSR, Moscow and Leningrad (1961), 398;
   b) S. K. Cherepanov, *Vascular Plants of Russia and Adjacent States (Within the Territory of the Former USSR)* [in Russian], Nauka, St. Petersburg (1995).
- 2. L. M. Belenovskaya, L. P. Markova, T. P. Nadezhina, and U. Ligaa, *Khim. Prir. Soedin.*, 575 (1977).
- 3. S. M. Adekenov, A. T. Kulyyasov, V. A. Raldugin, I. Yu. Bagryanskaya, Yu. V. Gatilov, and M. M. Shakirov, *Izv. Akad. Nauk, Ser. Khim.*, 167 (1998).
- 4. W. Z. Wang, R. X. Tan, Y. M. Yao, Y. Wang, and F. X. Jiang, *Phytochemistry*, 37, 1347 (1994).
- 5. E. E. Sistad, C. A. N. Catalan, A. B. Gutierrez, A. G. Diaz, V. L. Goedken, and W. Herz, *Phytochemistry*, **30**, 1933 (1991).
- 6. H. Li, J. C. Meng, C. H. K. Cheng, T. Higa, J. Tanaka, and R. X. Taan, J. Nat. Prod., 62, 1053 (1999).
- 7. M. A. Khanina, E. A. Serykh, G. A. Atazhanova, S. M. Adekenov, L. M. Pokrovskii, and A. V. Tkachev, *Khim. Rastit. Syr'ya*, 41 (1999).
- 8. a) F. Sorm, J. Novak, and V. Herout, *Collect. Czech. Chem. Commun.*, **18**, 527 (1953); b) R. A. Morton, ed., *Biochemical Spectroscopy*, Vol. 1, Adam Hilger, London (1975), p. 125.
- 9. M. Nakatsuji, Y. Hata, T. Fujihara, K. Yamamoto, M. Sasaki, H. Takekuma, M. Yoshihara, T. Minematsu, and S. Takekuma, *Tetrahedron*, **60**, 5983 (2004).
- 10. S. Takekuma, Y. Hata, T. Nishimoto, E. Nomura, M. Sasaki, T. Minematsu, and H. Takekuma, *Tetrahedron*, **61**, 6892 (2005).
- 11. R. V. Palei, V. V. Plemenkov, O. A. Lodochnikova, O. N. Kataeva, and I. A. Litvinov, *Zh. Org. Khim.*, **36**, 1772 (2000).
- 12. Y. Seo, J.-R. Rho, N. Geum, J. B. Yoon, and J. Shin, J. Nat. Prod., 59, 985 (1996).
- 13. J. Jakupovic, Z.-L. Chen, and F. Bohlmann, *Phytochemistry*, 26, 2777 (1987).